Binder Interactions in the Electrodes of the Lithium-ion Batteries

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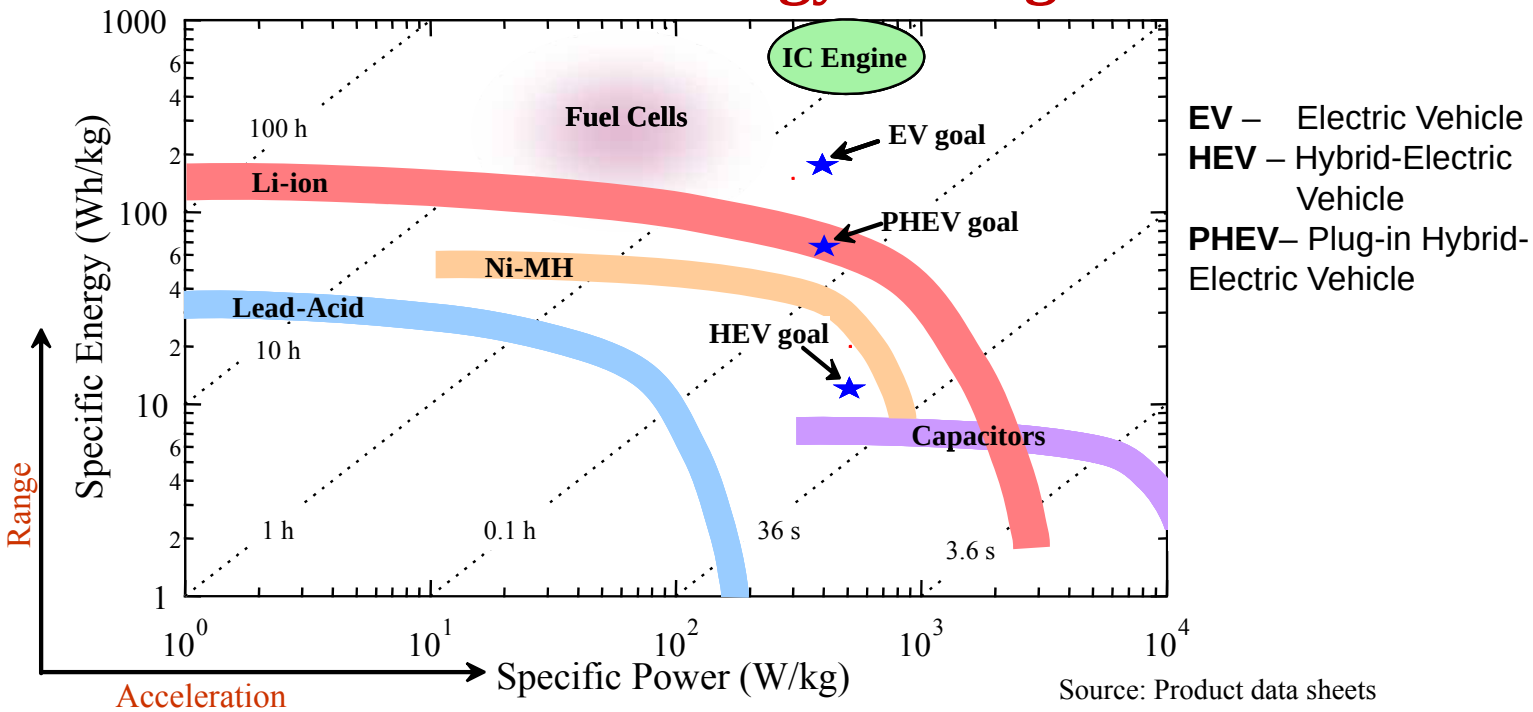
Outline

- A introduction to Lithium-ion battery.
  - Energy vs. Power.
  - Cost.
  - Structure.
  - Cathode and anode performance.
  - Cell performance improvement.

- Cathode electrode optimization.
  - Three components cathode – a polymer composite.
  - Acetylene black and polymer binder as conductive glue.
  - The mathematic modeling of the electrode electronic conductivity – the nano-scale interaction between PVDF and particles.
  - The cell performance governed by the interaction.

Relative Performance of Various Electrochemical Energy-Storage Devices



Source: Product data sheets

- Goals developed in cooperation with DOE and United States Advanced Battery Consortium (USABC) under the FreedomCAR partnership
  - USABC is a cooperative of major automotive manufactures

- Li-ion batteries have higher performance compared to Nickel-metal hydrides batteries. However, research is needed to simultaneously address the life, cost, and abuse tolerance issues

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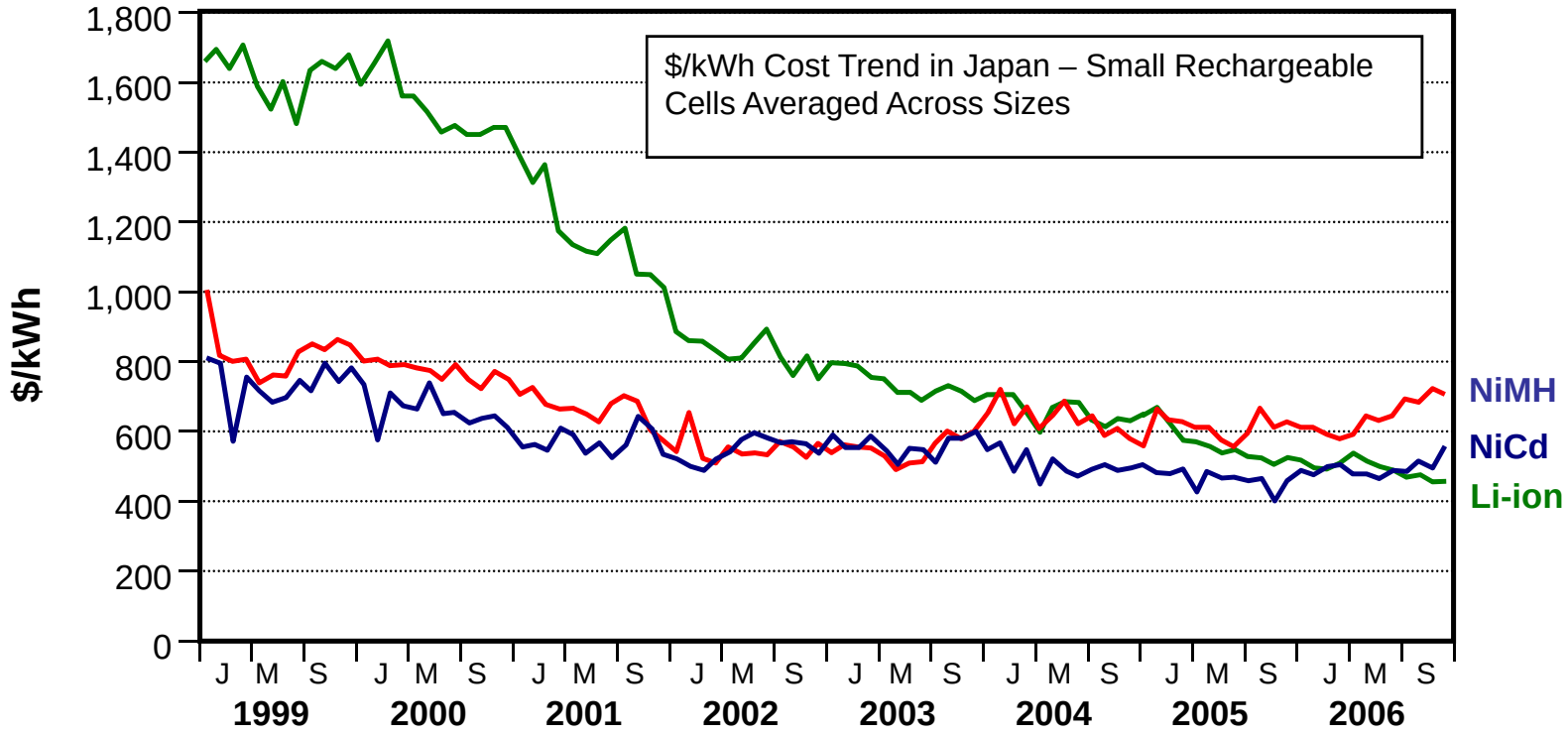
Cost of Consumer Electronics Batteries



Source: U.S. DOE

Source: TIAX, based on METI data

Performance and cost drivers for Li-ion cells
However, numerous problems remain before use in vehicles

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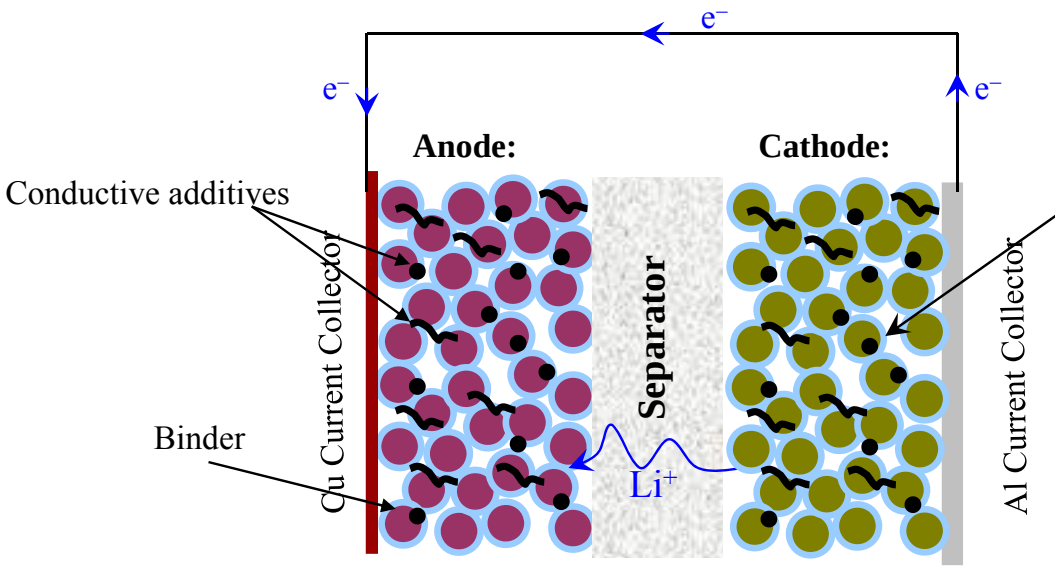
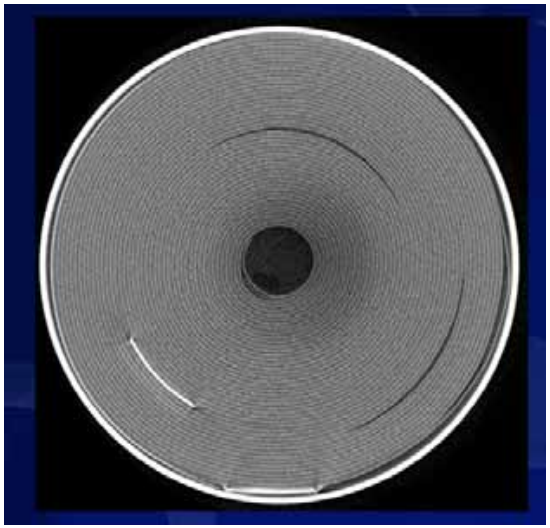
Modern Li-ion Battery

Lithium-ion battery



Electrolyte
$LiPF_6$ in Ethylene carbonate/diethyl carbonate



$Li^+ + e^- + C_6 \rightarrow LiC_6$

$LiCoO_2 \rightarrow Li^+ + e^- + CoO_2$

Innovation can occur via new material development, or by better engineering

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A commercial Lithium-ion Rechargeable 26650 Cell

Open from the anode side

Anode crimped closed

Cathode side of the jelly roll

Open up the jelly roll

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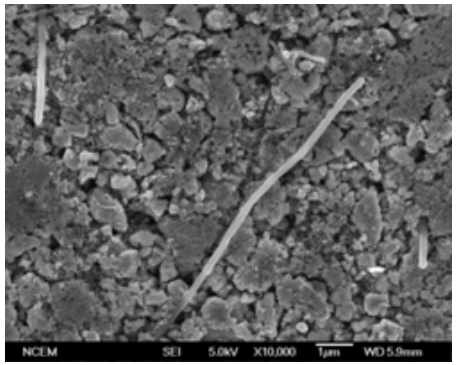
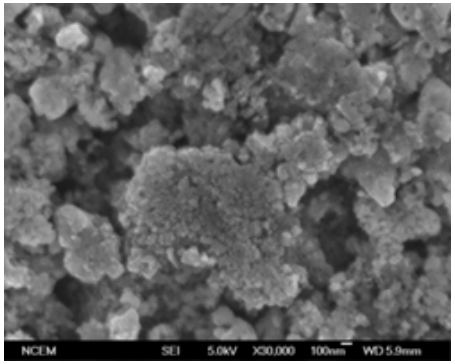
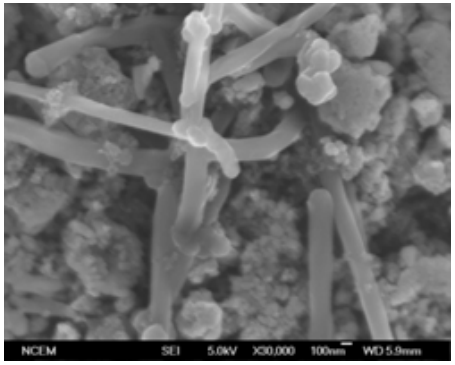
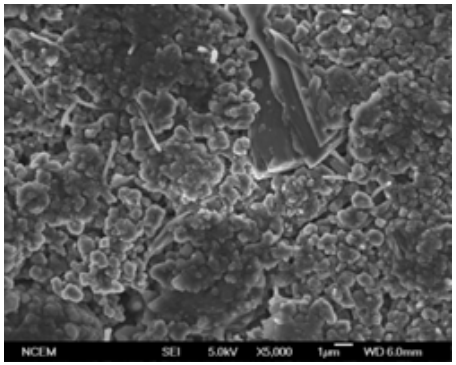
SEM Images of Cathode and Anode Surfaces

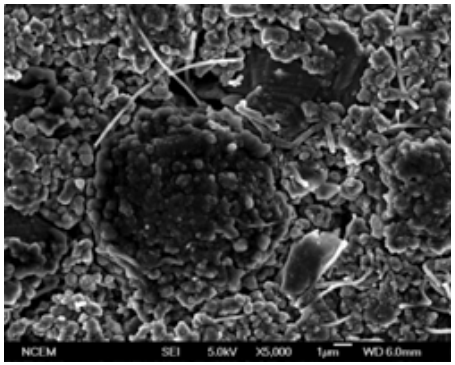


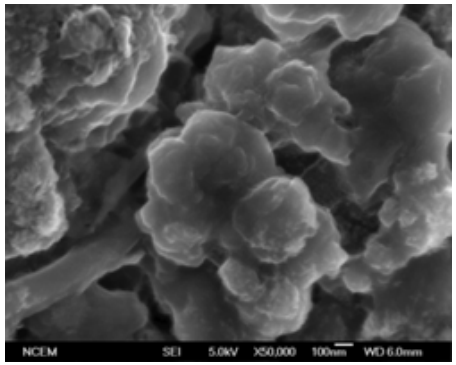




Cathode







Anode

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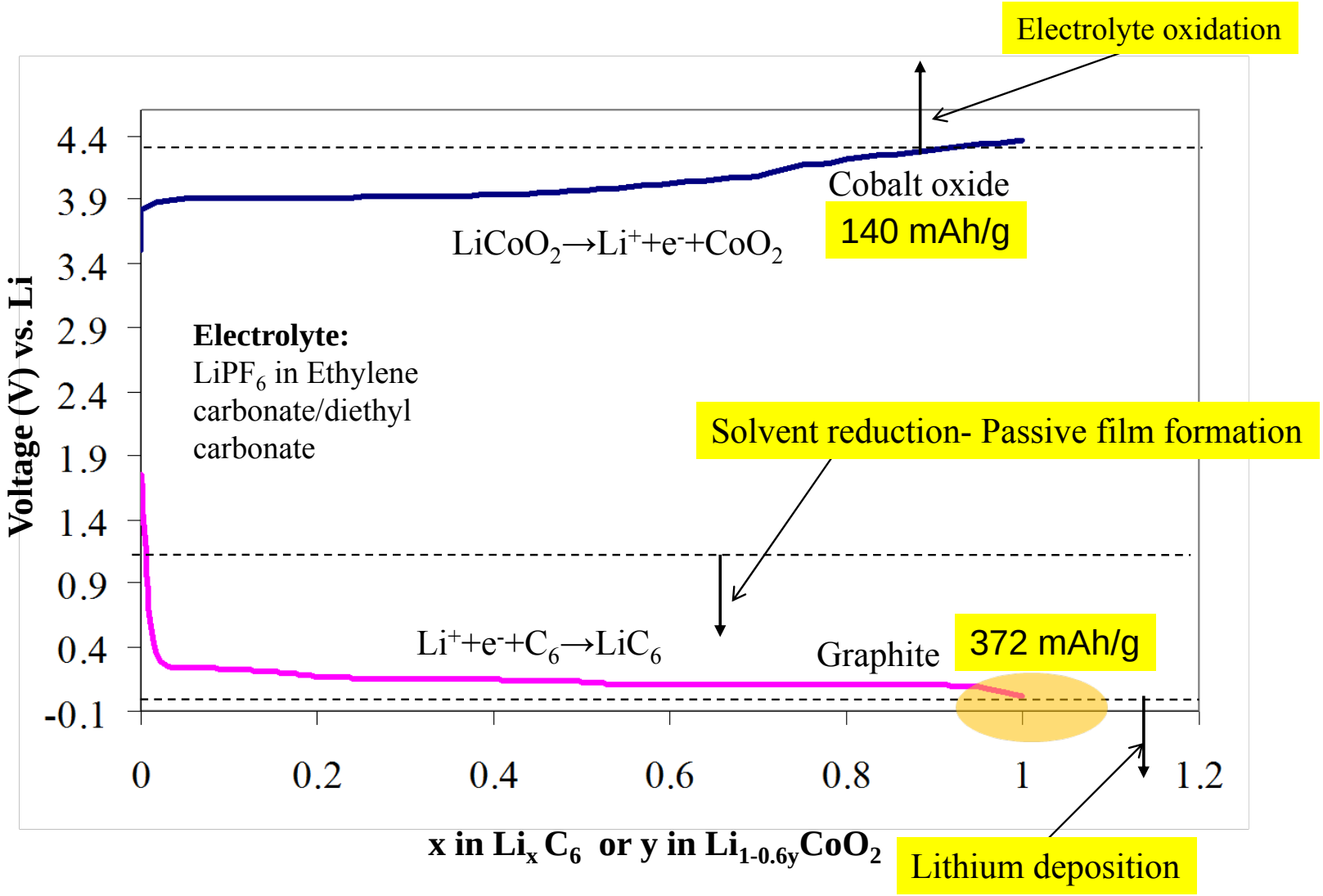
Cycling of a Graphite-$LiCoO_2$ Cell



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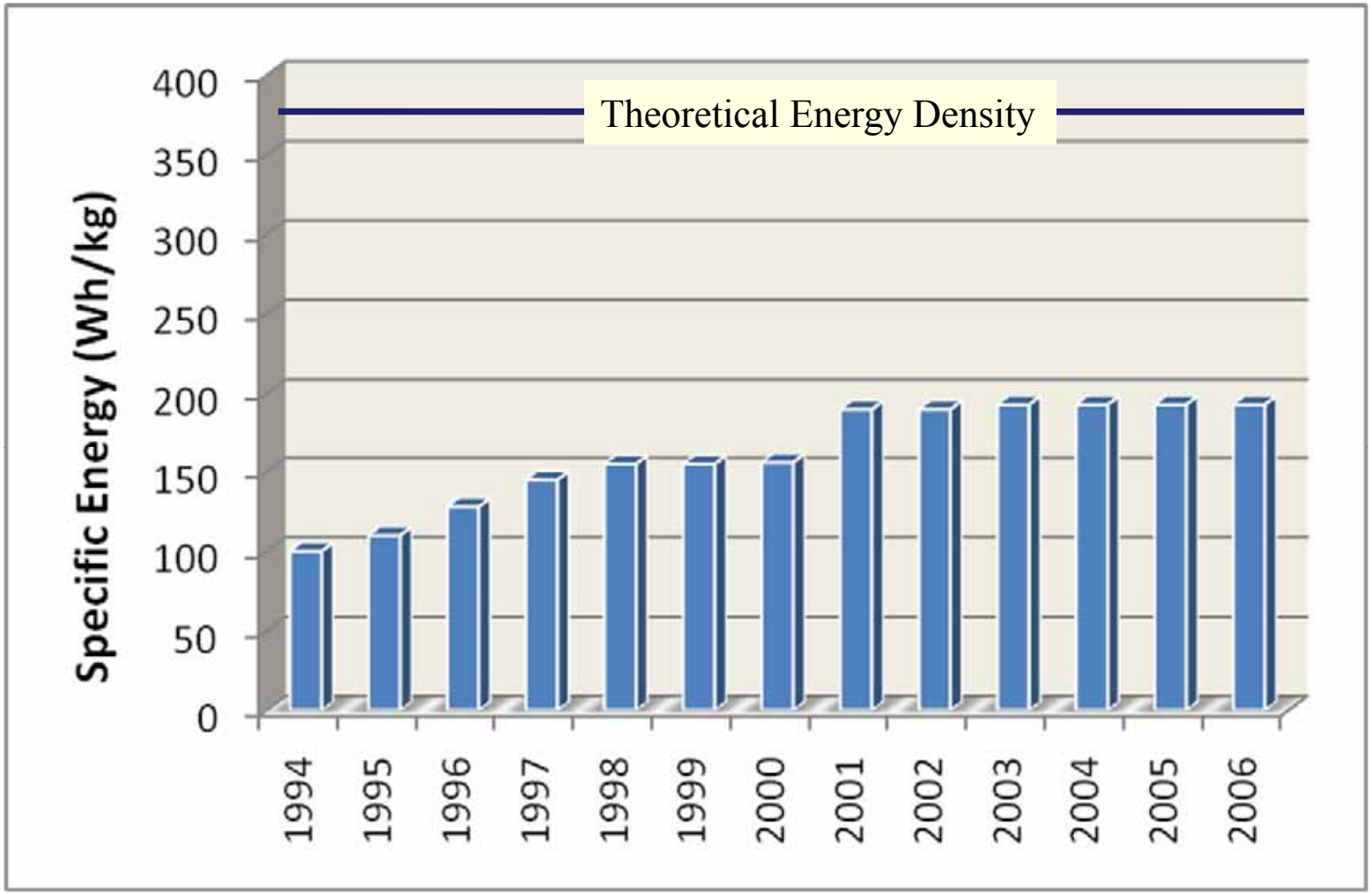
Energy Density Increase of Consumer Electronic Li-ion Batteries



Source: TIAX, LLC

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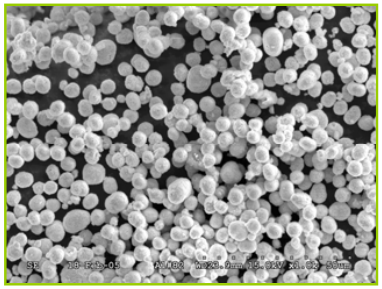
Complete Cathode Electrodes

Materials

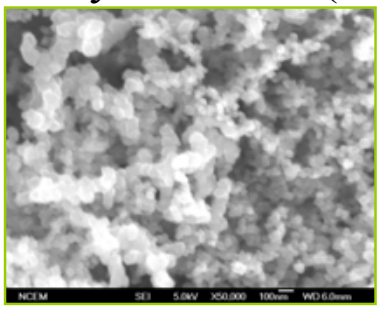
$LiNi_{0.80}Co_{0.15}Al_{0.05}O_2$



Ave. diam.: 10 μm
Surface area: 0.78 m^2/g

Scale bar: 10 μm

Acetylene black (AB)



Ave. diam.: 50 nm
Surface area: 60.4 m^2/g

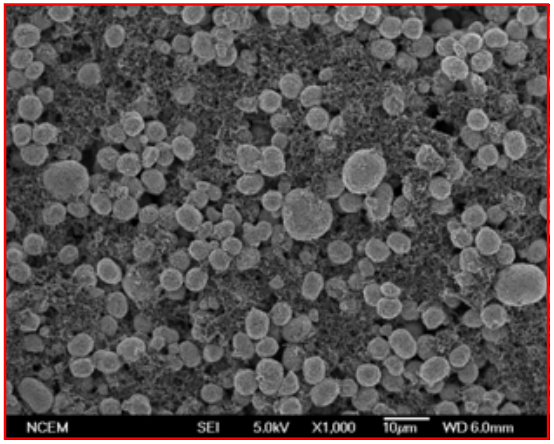
Scale bar: 100 nm

PVDF

Polymer Binder

Molecular weight: 760,000

Electrode



Scale bar: 10 μm

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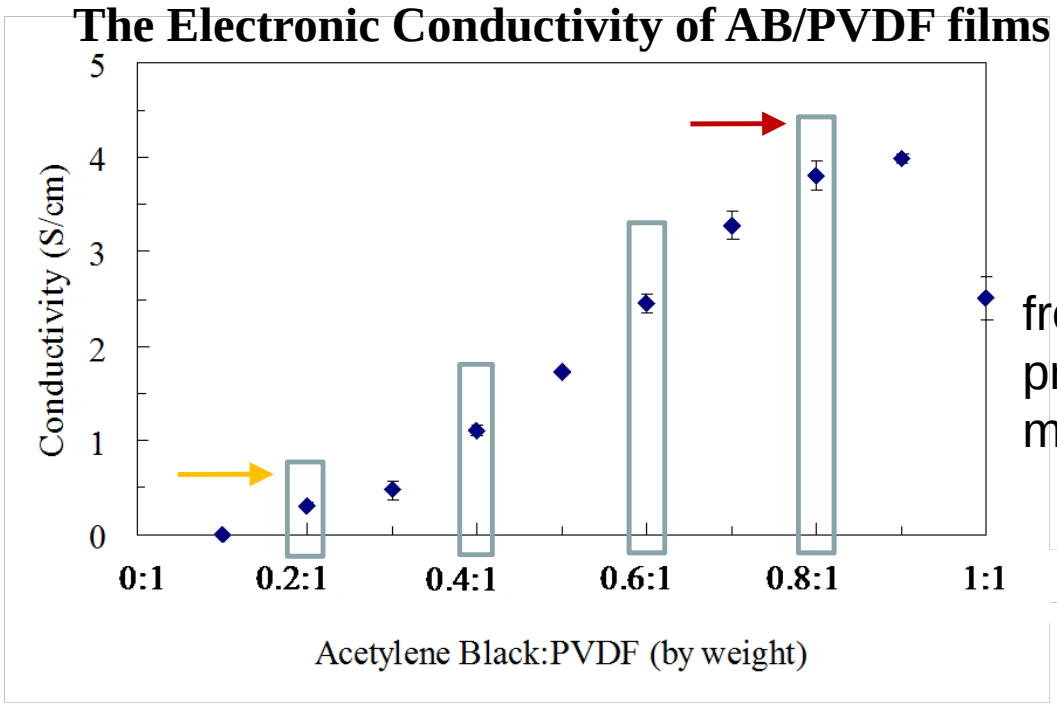
Laminates <u>without</u> Active Material



from 4-point probe measurements

$LiNi_{0.80}Co_{0.15}Al_{0.05}O_2$/Graphite Cells

	AB:PVDF	(AB+PVDF)% in the cathode electrode
Most conductive →	0.8:1	3.6%; 9%; 18%; 27%
Least conductive →	0.2:1	1.2%; 2.4%; 4.8%; 9.6%; 24%
	0.4:1	2.8%; 11.2%; 21%
	0.6:1	3.2%; 12.8%; 24%

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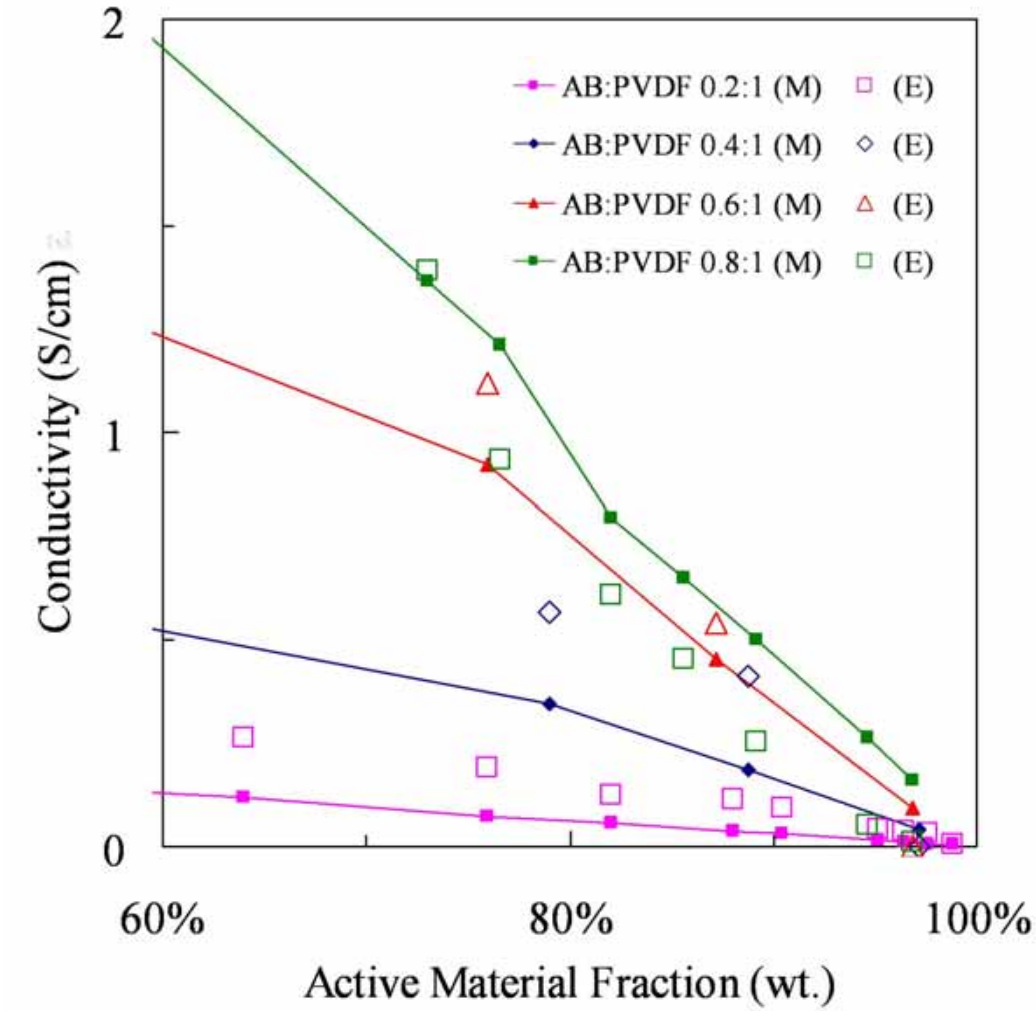
The Modeling Results of Active material/AB/PVDF

$$k_{eff} = \varepsilon^{\rho} * k_0(AB/PVDF)$$



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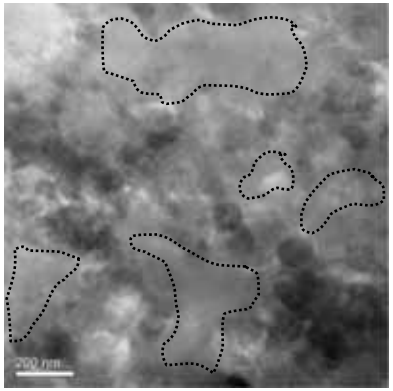
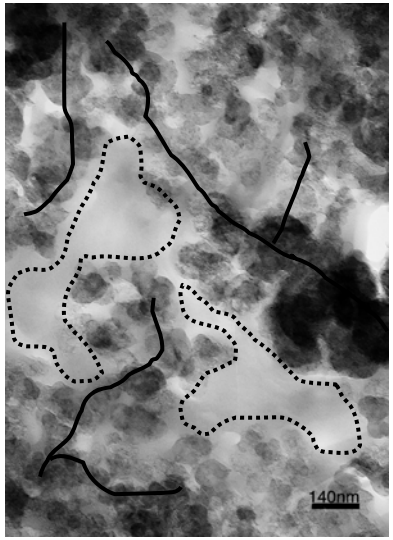
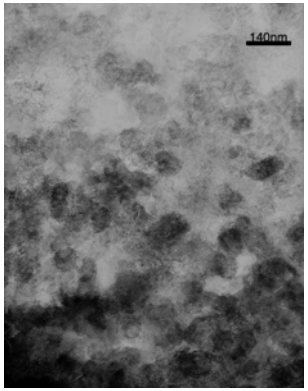
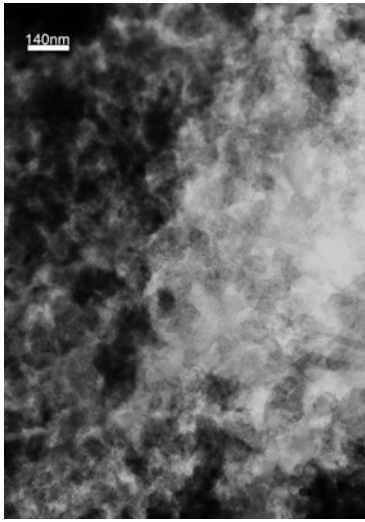
TEM Images of the Morphology of AB/PVDF Conductive Matrix

AB:PVDF = 0.2:1 (w/w) 0.5:1 0.8:1 1:1









200 nm 140 nm 140 nm 200 nm

Scale bar

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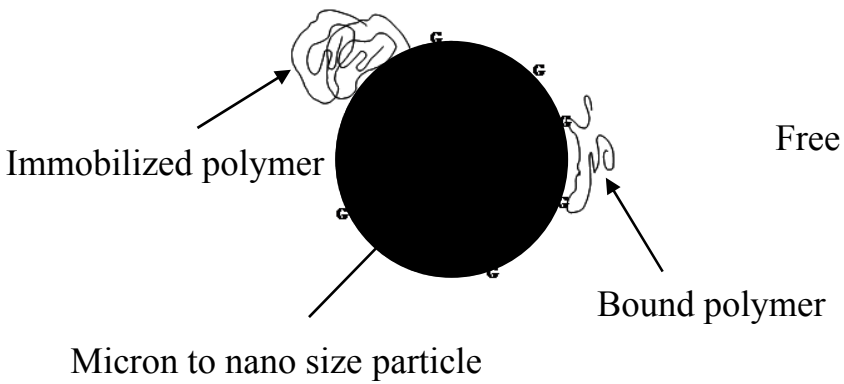
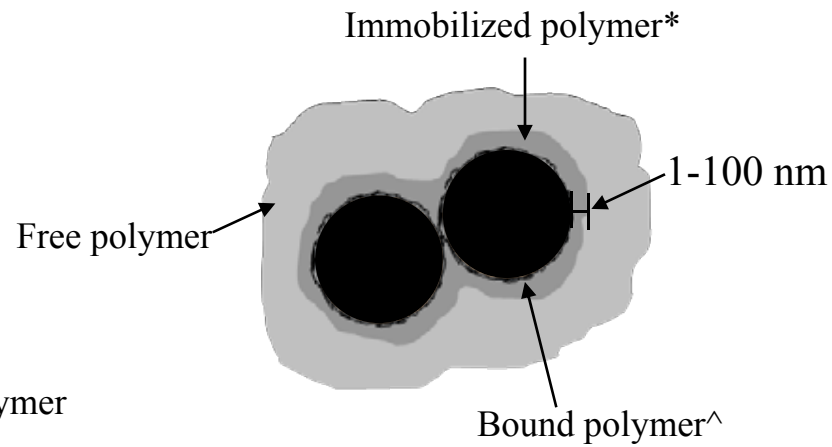
Three Different States of Polymer When in Contact with Particles





Bound polymer layer and immobilized polymer layer form Fixed Layer on the surface of particles.

* Dannenberg, E. M., Polymer, 14, 309 (1973)
^Westlinning, H and Butecnuth, G., Makromol. Chem., 47, 215 (1961)

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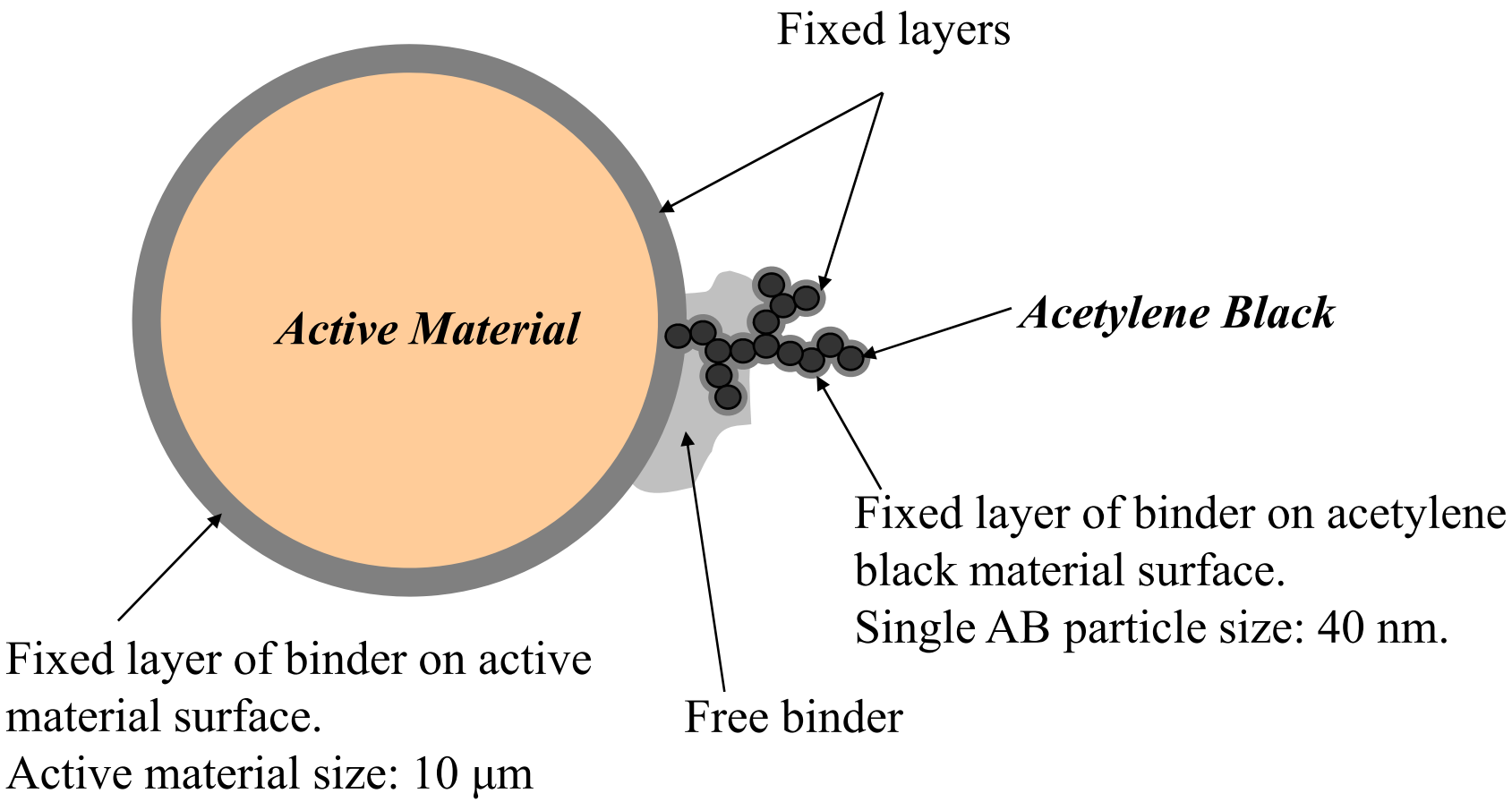
Cathode Material vs. Acetylene Black Particles



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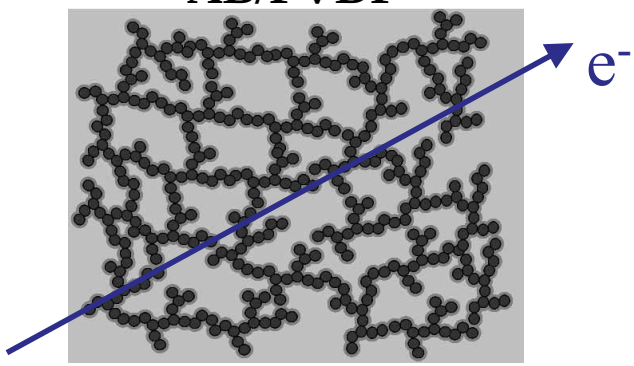
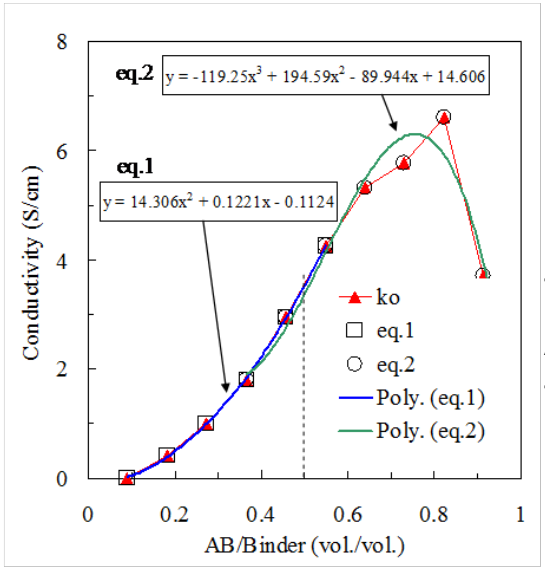
Matrix Conductivity of the Electrode Reflects the Competition for Binder between AB and Active Materials

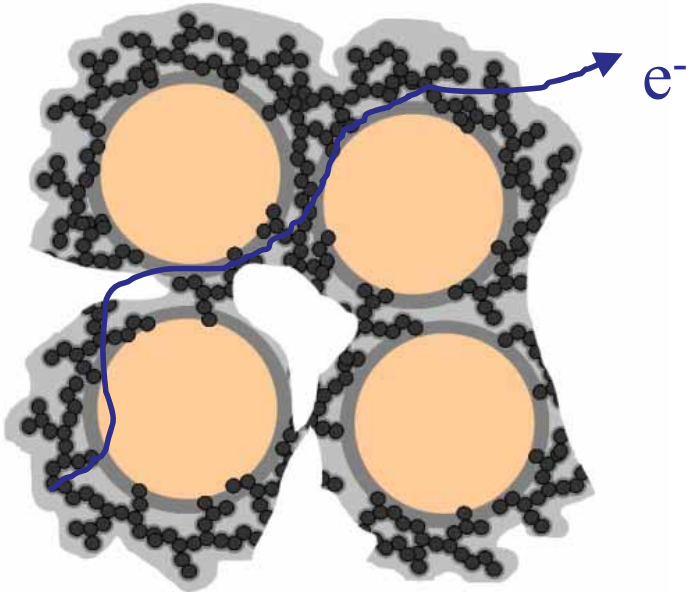
AB/PVDF





from AB/PVDF films

AB/PVDF/Active material



$$k_0 = f(AB/PVDF)$$

$$k_{eff} = \varepsilon^{\rho} * k_0(AB/PVDF')$$

PVDF' =PVDF - cathode particle immobilized PVDF

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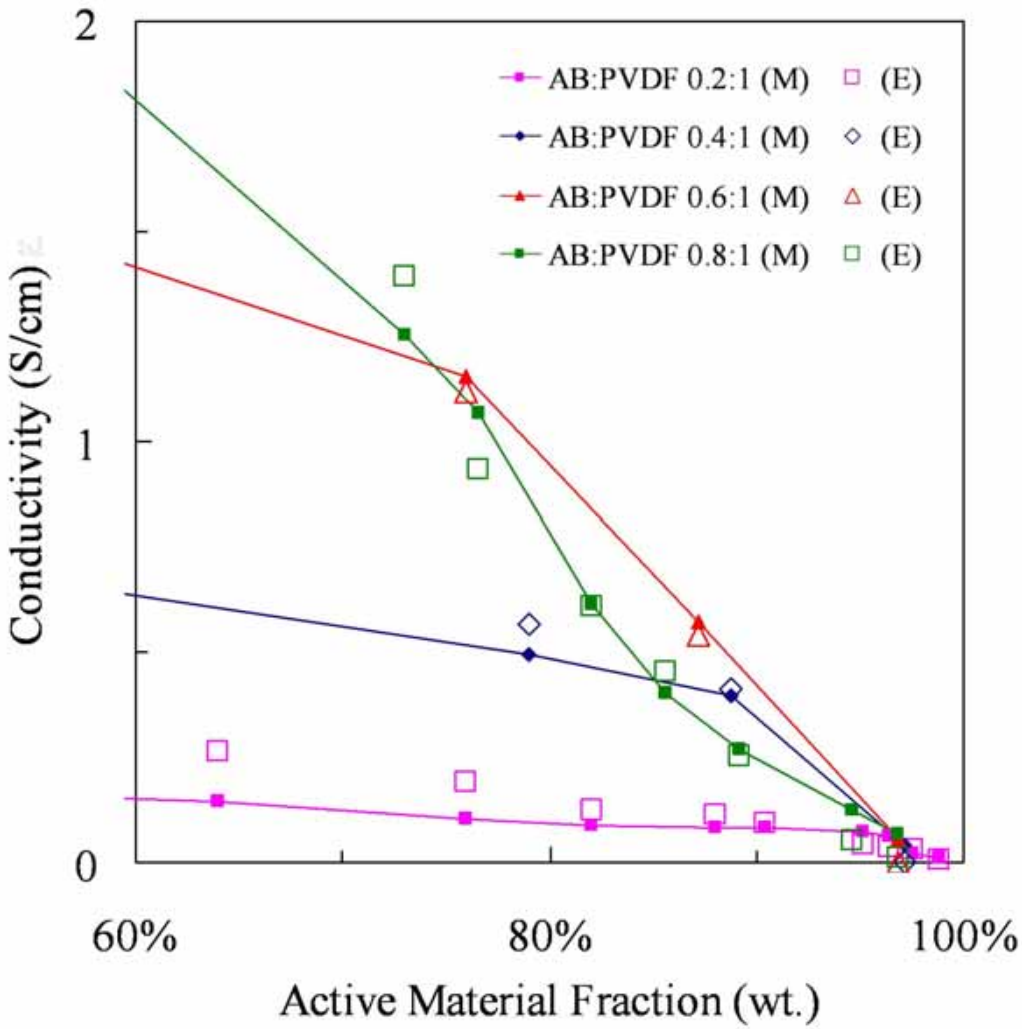
A Model and Experimental Comparison of the Conductivities Active Material/AB/PVDF



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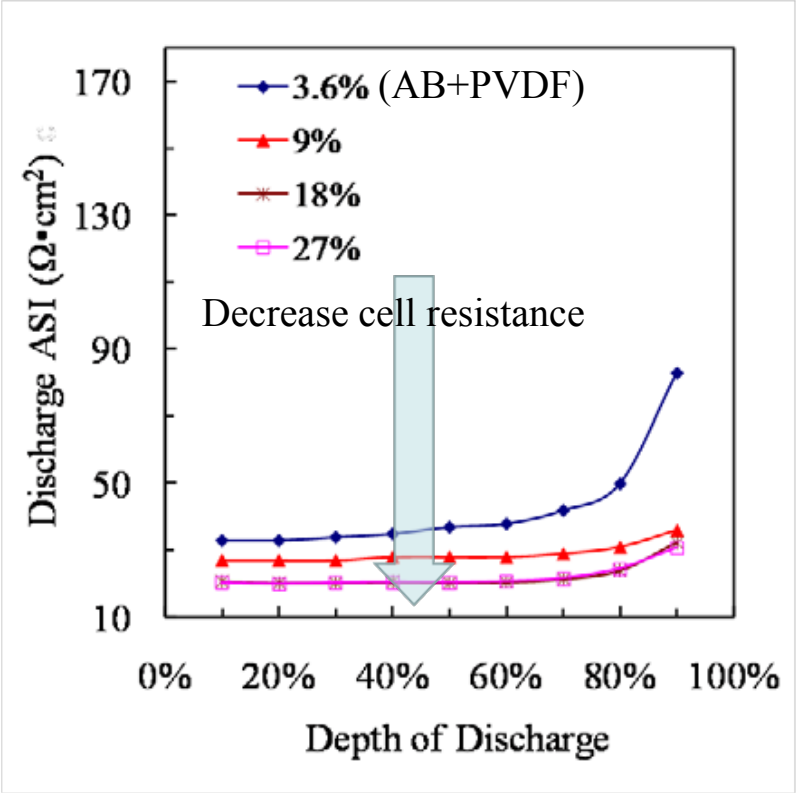
Complete Cells

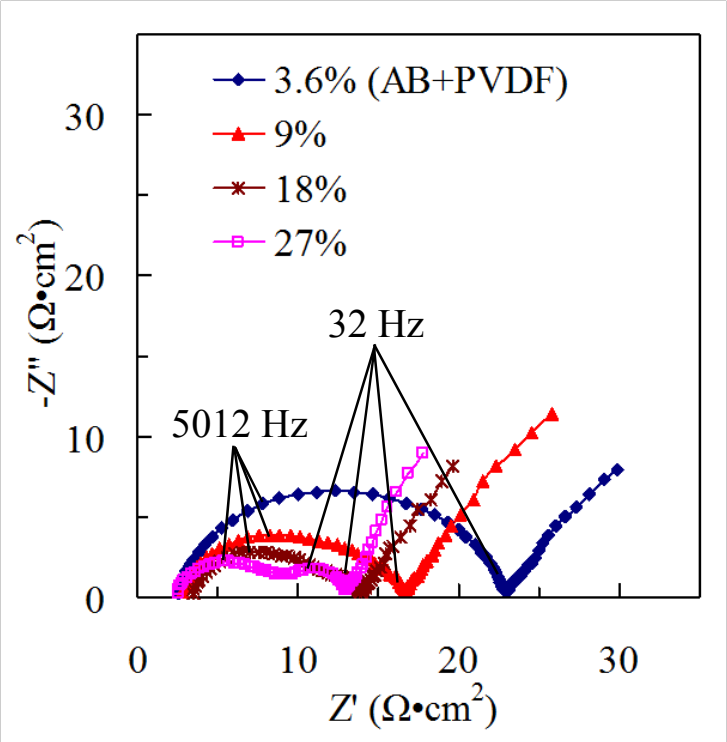
$LiNi_{0.80}Co_{0.15}Al_{0.05}O_2$/Graphite Cells

AB:PVDF = 0.8:1 (Most Conductive)

HPPC Test



EIS at 40% DOD

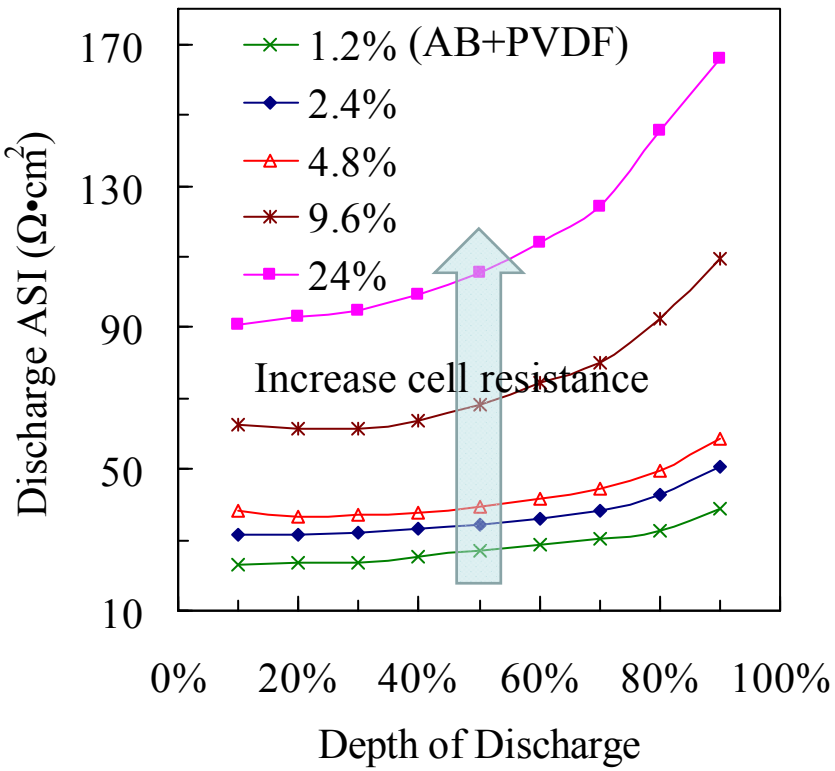


Duesenfeld Ex. 2006-018

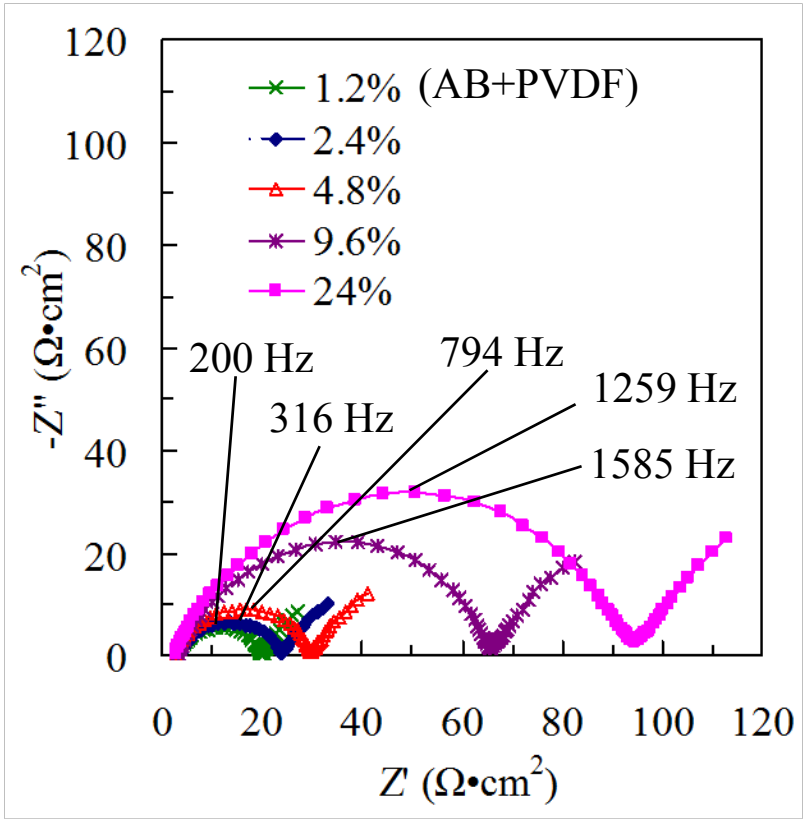
$LiNi_{0.80}Co_{0.15}Al_{0.05}O_2$/Graphite Cells

AB:PVDF = 0.2:1 (Least Conductive)

HPPC Test



EIS at 40% DOD

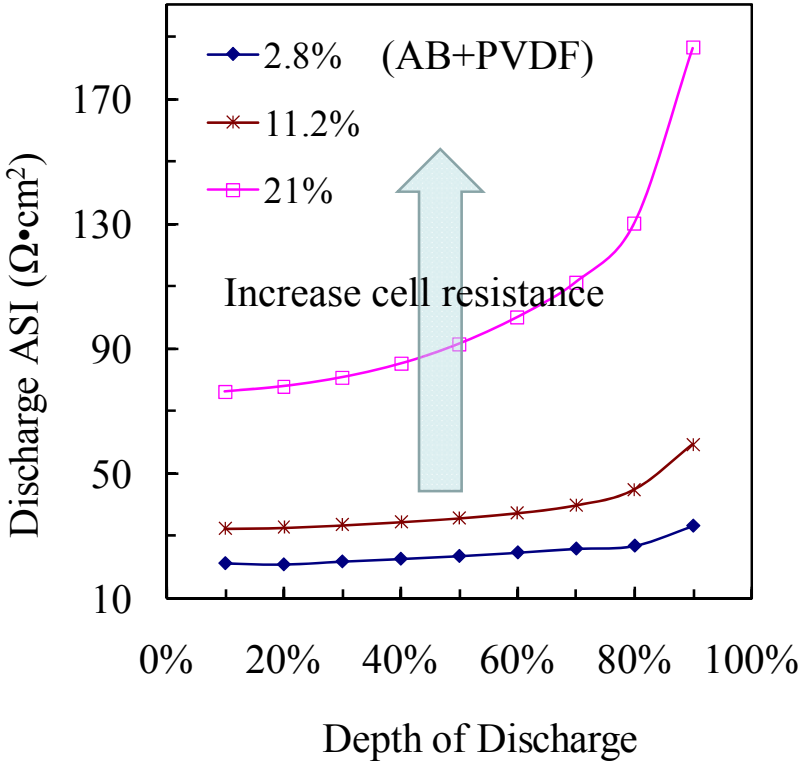


Duesenfeld Ex. 2006-019

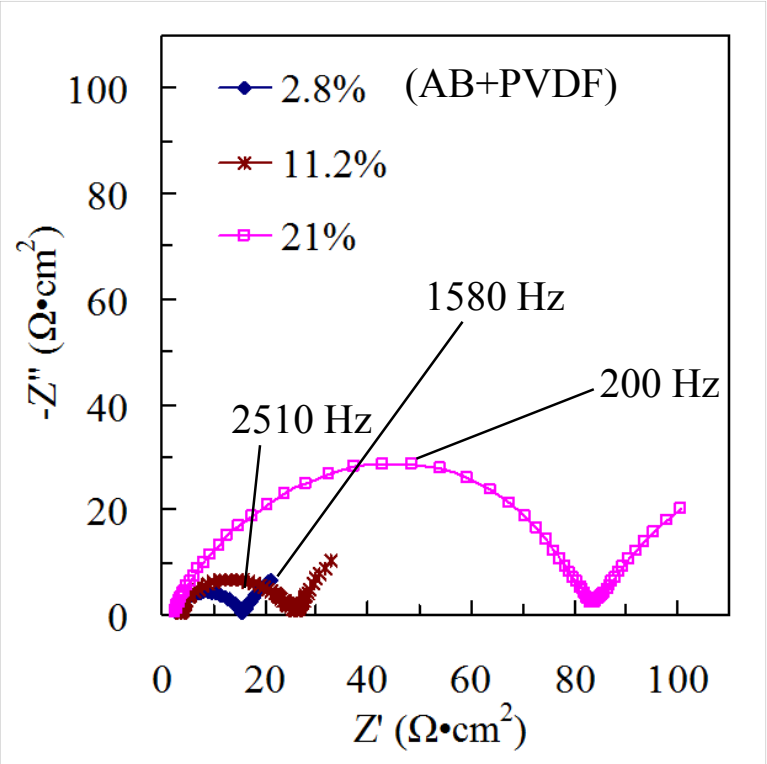
$LiNi_{0.80}Co_{0.15}Al_{0.05}O_2$/Graphite Cells

AB:PVDF = 0.4:1

HPPC Test



EIS at 40% DOD



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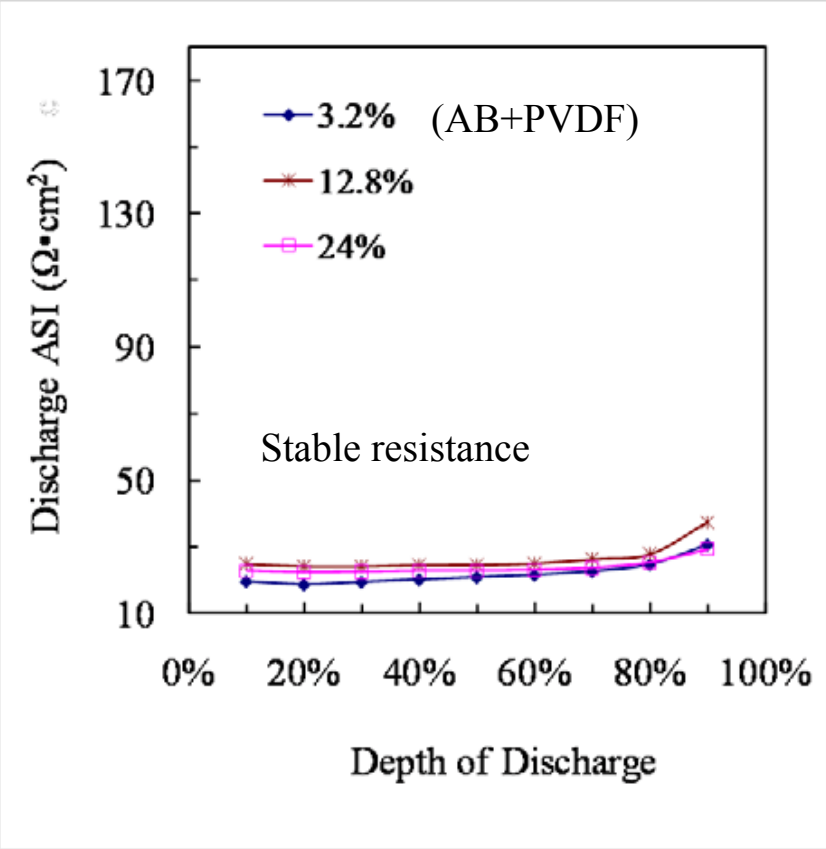
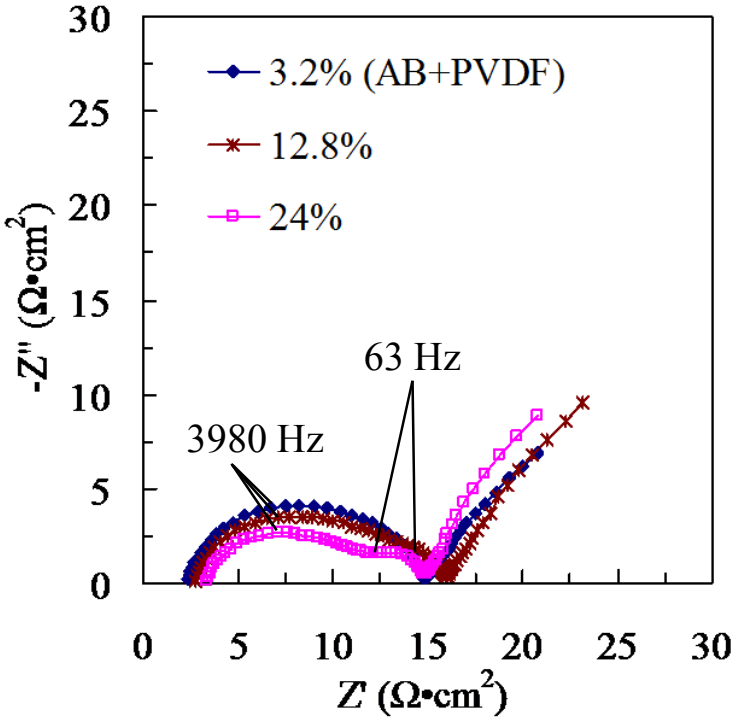
$LiNi_{0.80}Co_{0.15}Al_{0.05}O_2$/Graphite Cells

AB:PVDF = 0.6:1

HPPC Test



EIS at 40% DOD



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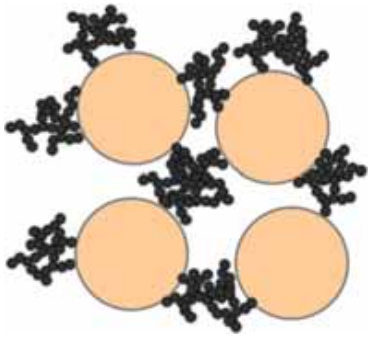
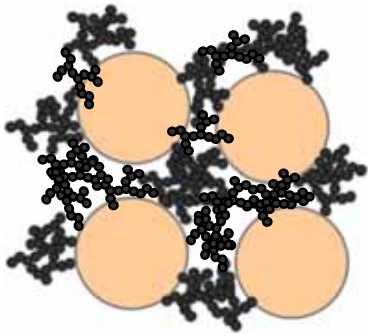
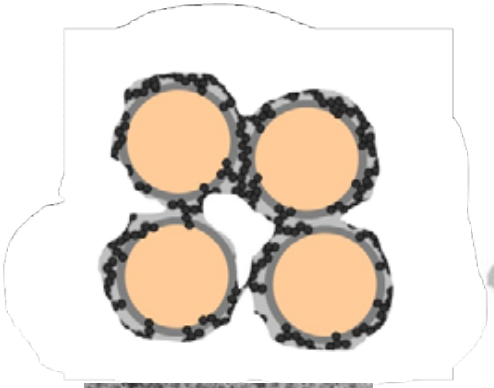
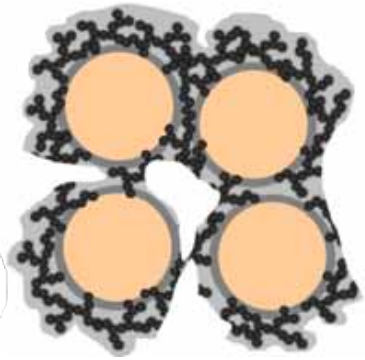
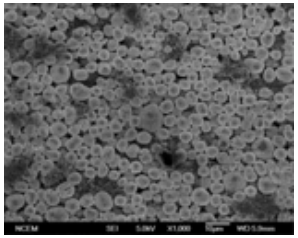
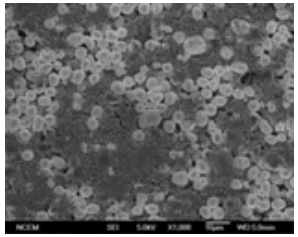
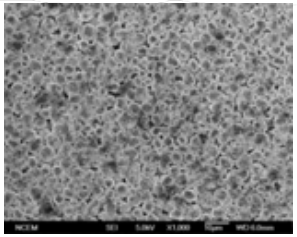
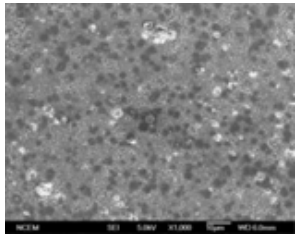
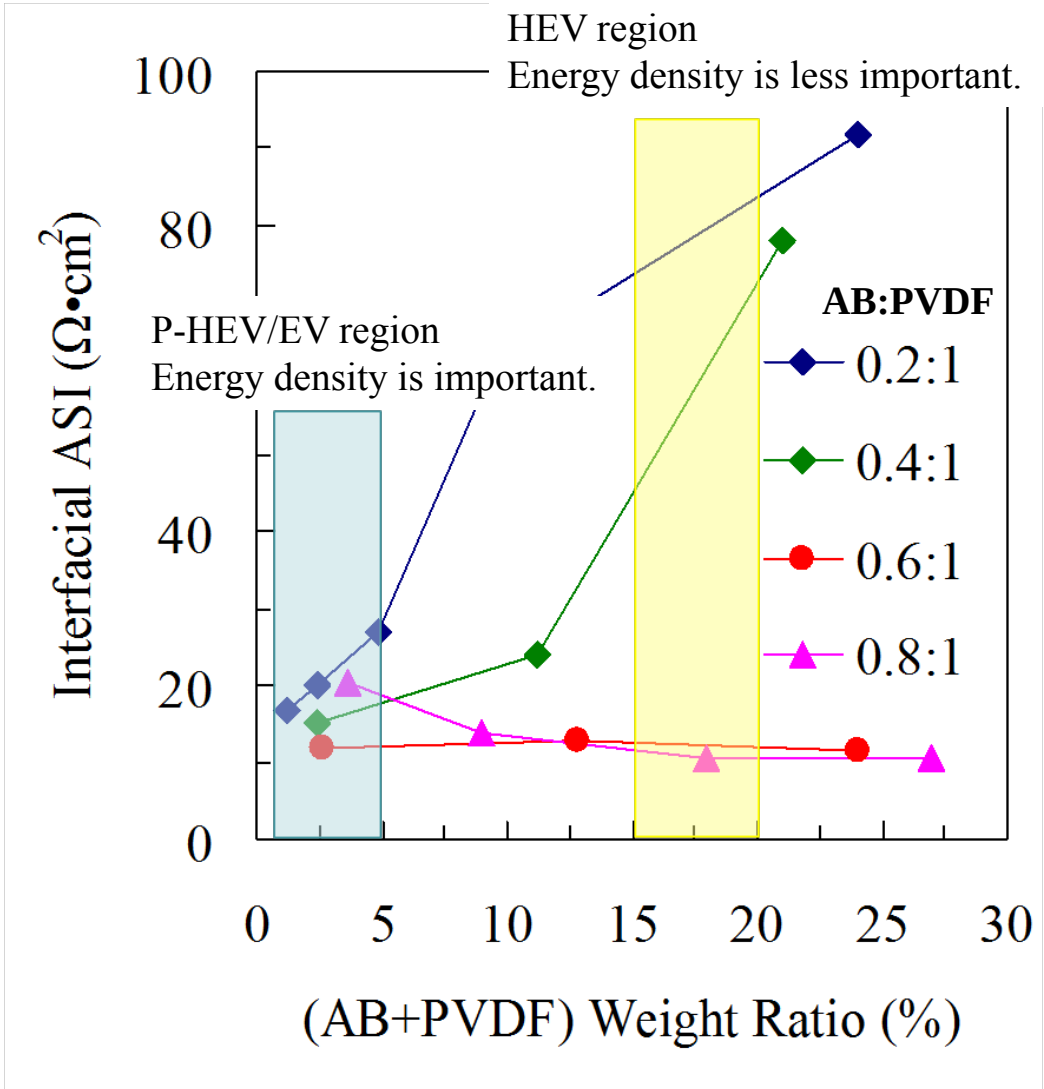
The Distribution of Binder, AB, and Active Material at Different Combinations

Active Material Content:	AB:PVDF = 0.8:1 High	AB:PVDF = 0.8:1 Low	AB:PVDF = 0.2:1 High	AB:PVDF = 0.2:1 Low
				
				
Electronic Conductivity:	↓	**Small Change**	↑	**Small Change**
Interfacial impedance of the electrodes:	↑	↓	↓	↑

Duesenfeld Ex. 2006-022

The Interfacial Impedances of All Compositions



Duesenfeld Ex. 2006-023

Conclusions

•The amount of binder plays critical role in the composite electrode beyond providing mechanical integrity.

•The binder to conductive carbon ratio dictates the electronic conductivity of the laminate. The addition of active material alters the ratio through competition for the binder.

•The electronic conductivity of a laminate appears to be reflected in the charge transfer capability of the laminate.

•High acetylene black content, such as AB:PVDF > 0.8:1, tends to produce electrodes with lower electronic conductivity and higher charge transfer at high active material loadings.

•Low acetylene black content, such as AB:PVDF < 0.4:1, tends to produce electrodes with higher electronic conductivities at high active material loadings.

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Thank you for your attention!

End